



2009 Ozone Summary

New Jersey Department of Environmental Protection

NATURE AND SOURCES

Ozone (O_3) is a gas consisting of three oxygen atoms. It occurs naturally in the upper atmosphere (stratospheric ozone) where it protects us from harmful ultraviolet rays (see Figure 1). However, at ground-level (tropospheric ozone) it is considered an air pollutant and can have serious adverse health effects. Ground-level ozone is created when nitrogen oxides (NO_x) and volatile organic compounds (VOC's) react in the presence of sunlight and heat. NO_x is primarily emitted by motor vehicles, power plants, and other sources of combustion. VOC's are emitted from sources such as motor vehicles, chemical plants, factories, consumer and commercial products, and even natural sources such as trees. Ozone and the pollutants that form ozone (precursor pollutants) can also be transported into an area from sources hundreds of miles upwind.

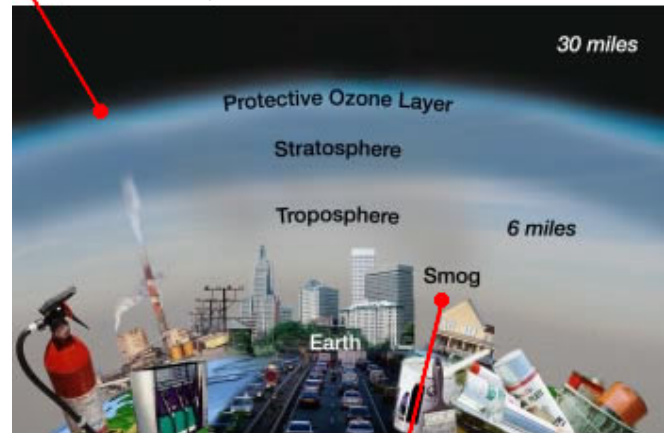
Since ground-level ozone needs sunlight to form, it is mainly a daytime problem during the summer months. Weather patterns have a significant effect on ozone formation and hot, dry summers will result in more ozone than cool, wet ones. In New Jersey, the ozone monitoring season runs from April 1st to October 31st. For a more complete explanation of the difference between ozone in the upper and lower atmosphere, see the U.S. Environmental Protection Agency (EPA) publication "Ozone: Good Up High, Bad Nearby".

ENVIRONMENTAL EFFECTS

Ground-level ozone damages plant life and is responsible for 500 million dollars in reduced crop production in the United States each year. It interferes with the ability of plants to produce and store food, making them more susceptible to disease, insects, other pollutants, and harsh weather. "Bad" ozone damages the foliage of trees and other plants, sometimes marring the landscape of cities, national parks and forests, and recreation areas. The black areas on the leaves of the blackberry bush and sassafras tree shown in Figure 2 and Figure 3 is damage caused by exposure to ground-level ozone. (Figure 2 and 3 Photos by: Teague Prichard, Wisconsin Department of Natural Resources)

Figure 1: Good and Bad Ozone

Ozone is good up here... Many popular consumer products like air conditioners and refrigerators involve CFCs or halons during either manufacturing or use. Over time, these chemicals damage the earth's protective ozone layer.



Ozone is bad down here... Cars, trucks, power plants and factories all emit air pollution that forms ground-level ozone, a primary component of smog.

Source: EPA

Figure 2: Ozone Damage to Blackberry Bush



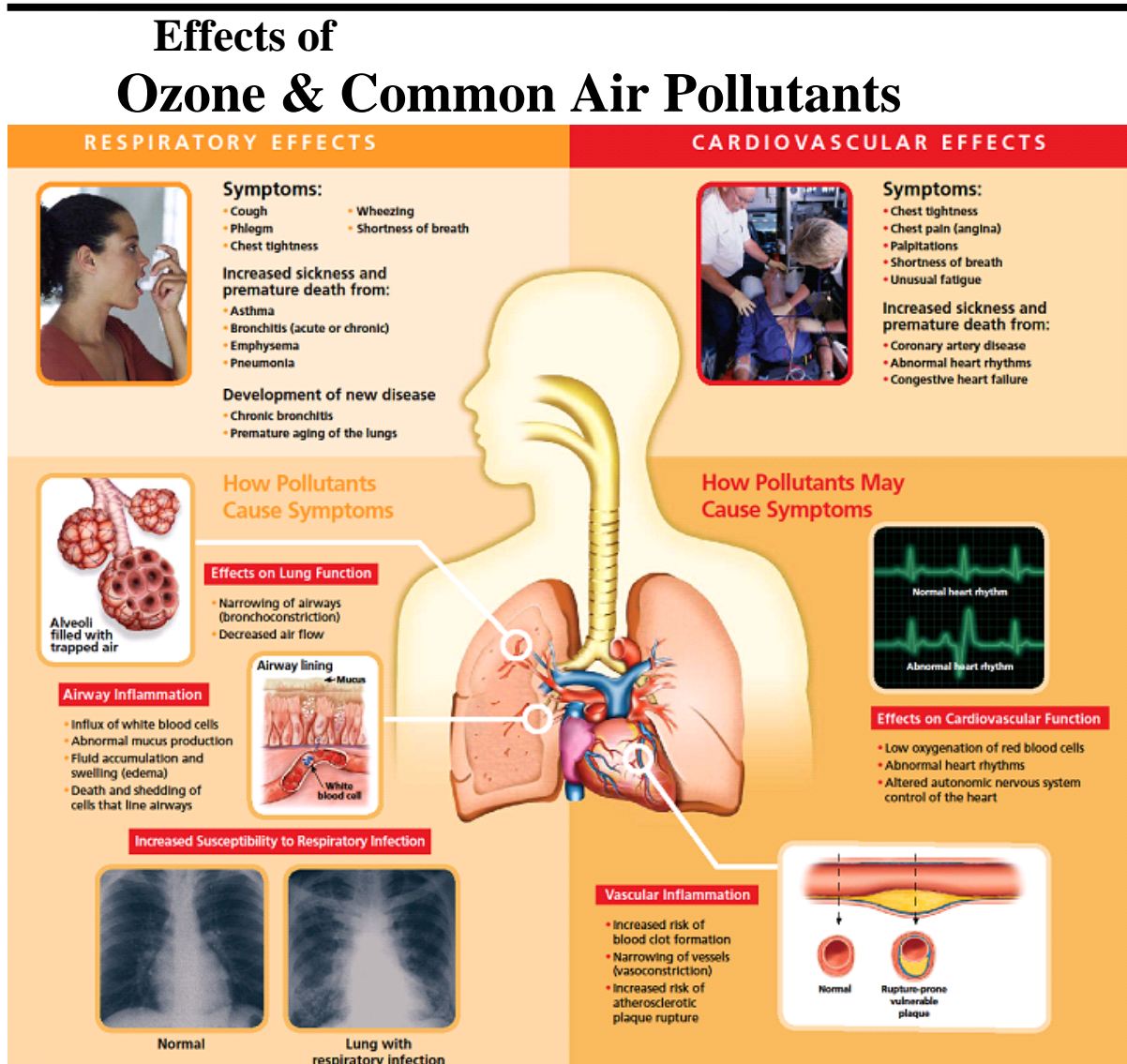
Figure 3: Ozone Damage to Sassafras Tree



HEALTH EFFECTS

Repeated exposure to ozone pollution may cause permanent damage to the lungs. Even when ozone is present in low levels, inhaling it can trigger a variety of health problems including chest pains, coughing, nausea, throat irritation, and congestion. Ozone also can aggravate other health problems such as bronchitis, heart disease, emphysema, and asthma, and can reduce lung capacity. People with pre-existing respiratory ailments are especially prone to the effects of ozone. For example, asthmatics affected by ozone may have more frequent or severe attacks during periods when ozone levels are high. As shown in Figure 4 ozone can irritate the entire respiratory tract. Children are also at risk for ozone related problems. Their respiratory systems are still developing and they breathe more air per pound of body weight than adults. They are also generally active outdoors during the summer when ozone levels are at their highest. Anyone who spends time outdoors in the summer can be affected and studies have shown that even healthy adults can experience difficulty in breathing when exposed to ozone. Anyone engaged in strenuous outdoor activities, such as jogging, should limit activity to the early morning or late evening hours on days when ozone levels are expected to be high.

Figure 4



Source: www.airnow.gov

AMBIENT AIR QUALITY STANDARDS FOR OZONE

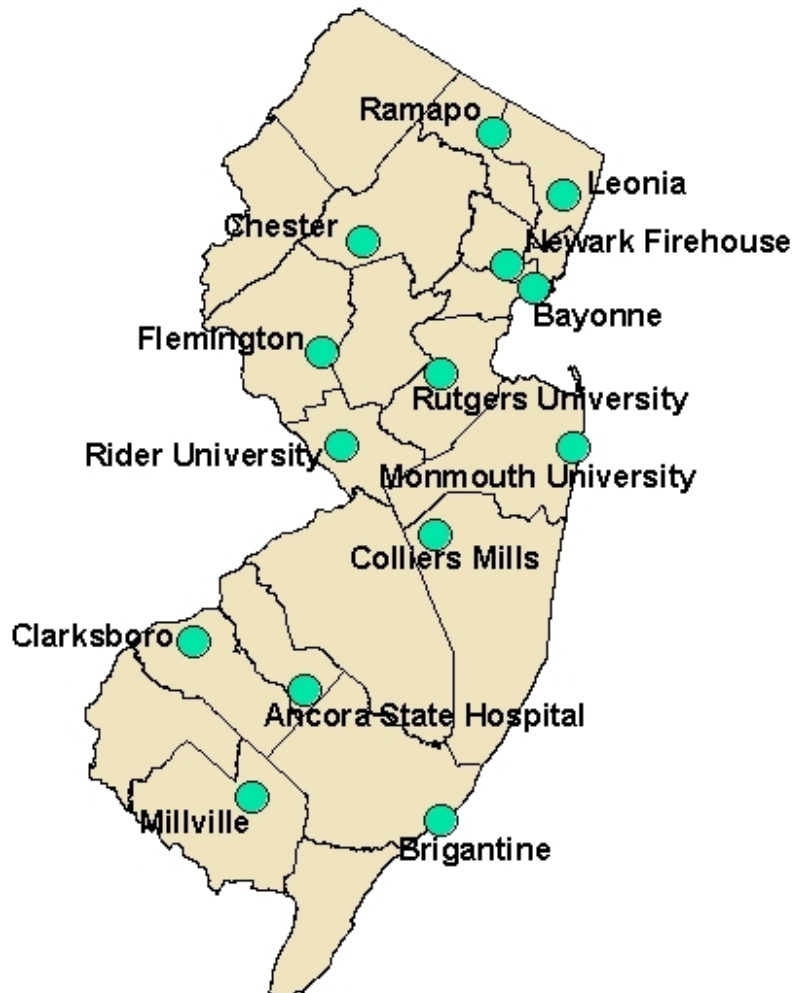
National and state air quality standards have been established for ground-level ozone. There are both primary standards, which are based on health effects, and secondary standards, which are based on welfare effects (e.g. damage to trees, crops and materials). For ground-level ozone, the primary and secondary National Ambient Air Quality Standards (NAAQS) are the same (see Table 1). The ozone NAAQS were revised in 2008 because EPA determined that the old standard of 0.08 parts per million (ppm) maximum daily eight-hour average was not sufficiently protective of public health. The revised standard of 0.075 ppm maximum daily 8-hour average went into effect on May 27, 2008. As many people are accustomed to the old standards, summary information relative to that standard will be provided in this report along with summaries based on the new standard.

Table 1
National and New Jersey Ambient Air Quality Standards for Ozone

ppm = Parts per Million

Averaging Period	Type	New Jersey	National
1-Hour	Primary	0.12 ppm	-----
1-Hour	Secondary	0.08 ppm	-----
8-Hour	Primary	-----	0.075 ppm
8-Hour	Secondary	-----	0.075 ppm

Figure 5
2009 Ozone Monitoring Network



OZONE NETWORK

Ozone was monitored at 14 locations in New Jersey during 2009. Of those 14 sites, 11 operated year round and 3 operated only during the ozone season (April 1st through October 31st). Colliers Mills, Monmouth University, and Ramapo were only operated during the ozone season.

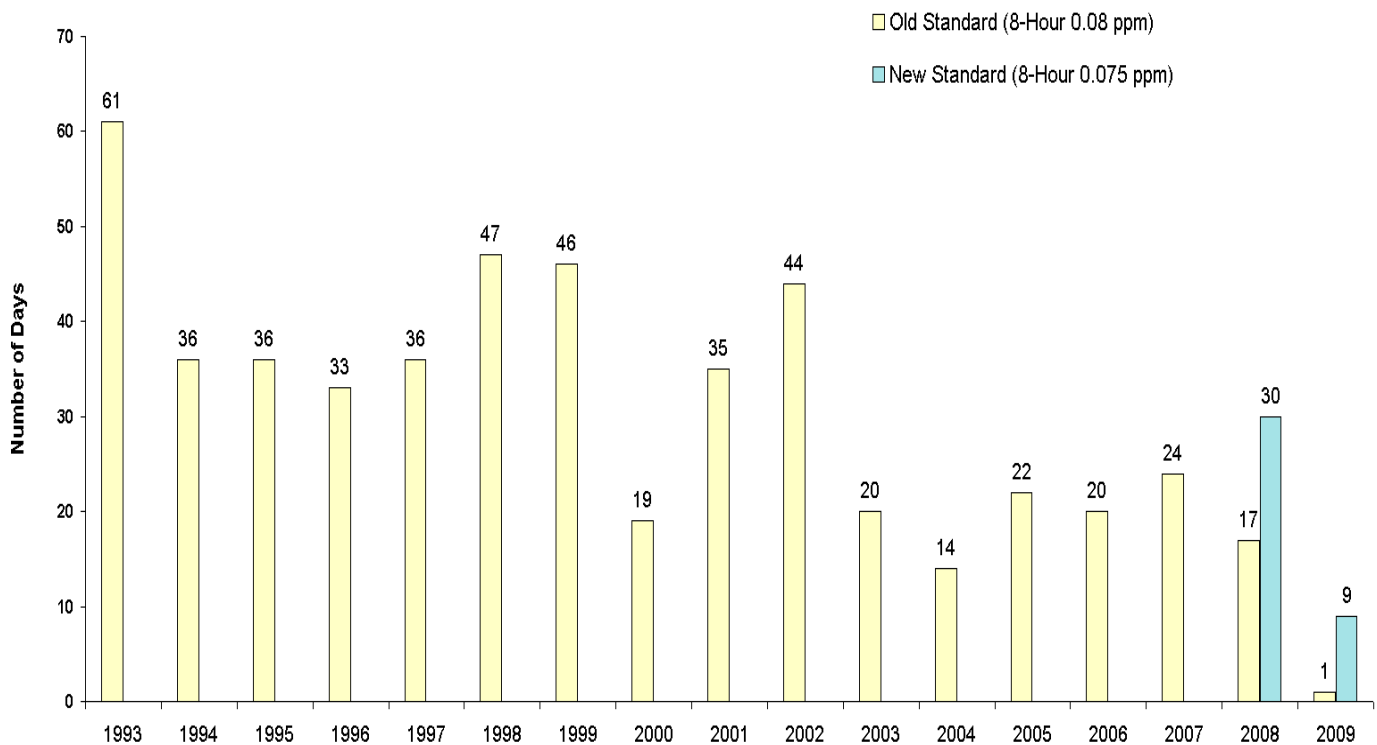
Site locations are shown in Figure 5.

HOW THE CHANGES TO THE OZONE STANDARDS AFFECT AIR QUALITY RATINGS

In regards to ground level ozone pollution, 2009 was the cleanest year on record. Just nine days exceeded the 0.075 ppm 8-hour standard and just one day exceeded the old 0.08 ppm 8-hour standard. There are fewer days on which those old standards are exceeded, and when they are, fewer sites tend to be involved. Also, the maximum levels reached are not as high as they were in the past. The maximum 1-hour average concentration recorded in 1993 was 0.162 ppm, compared to a maximum of 0.109 ppm in 2009.

It is apparent, however, that the current standard is significantly more stringent than the old (see Figure 6 below). As a result, additional control measures to reduce ozone levels will be needed. These measures will have to be implemented over a wide area and will require the cooperative effort of many states and the federal government if they are to be successful.

Figure 6
Days on Which the Old and New
Ozone Standards Have Been Exceeded in New Jersey
1993 - 2009



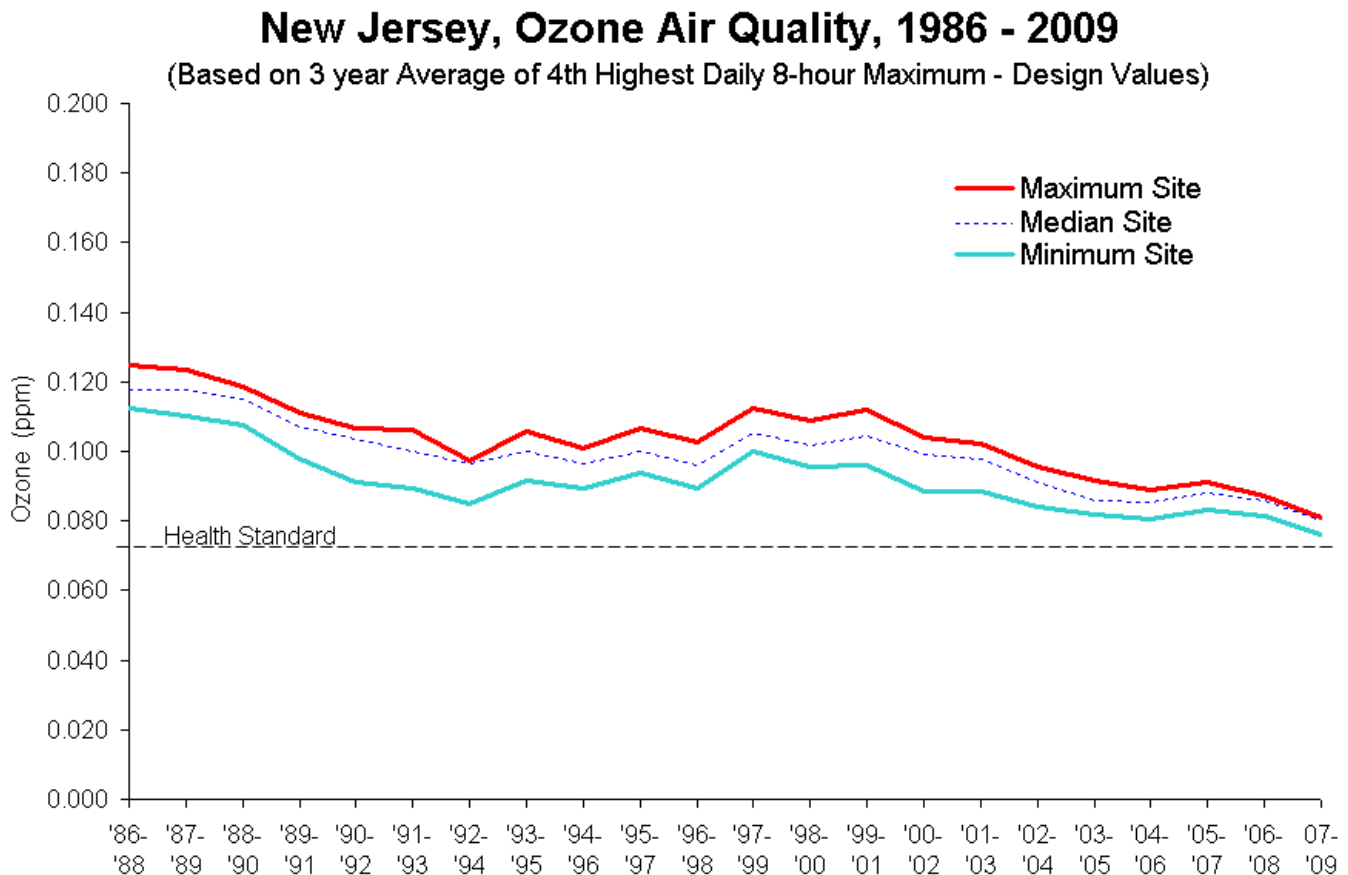
DESIGN VALUES

The NAAQS for ozone are set in such a way that determining whether they are being attained is not based on a single year. For example, an area was considered to be attaining the old 1-hour average standard if the average number of times the standard was exceeded over a three-year period was 1 or less (after correcting for missing data). Thus it was the fourth highest daily maximum 1-hour concentration that occurred over a three-year period that determined if an area would be in attainment. If the fourth highest value was above 0.12 ppm then the average number of exceedances would be greater than 1. The fourth highest value is also known as the design value.

Under the new standard, attainment is determined by taking the average of the fourth highest daily maximum 8-hour average concentration that is recorded each year for three years. This becomes the design value for an area under the new standard. When plans are developed for reducing ozone concentrations, an area must demonstrate that the ozone reduction achieved will be sufficient to ensure the design value will be below the NAAQS, as opposed to ensuring that the standards are never exceeded. This avoids developing plans based on extremely rare events.

Figure 7 shows the design value for the 8-hour standards starting with the 1986-1988 period. Design values are calculated for all ozone sites in the network and the median, maximum, and minimum for each year were used in the graphics.

Figure 7



SUMMARY OF 2009 Ozone Data Relative to the OLD 1-HOUR STANDARD

Of the 14 monitoring sites that were operated during the 2009 ozone season, none recorded levels above the old 1-hour standard of 0.12 ppm during the year. The highest 1-hour concentration was 0.109 ppm recorded at Colliers Mills on August 17th. This is the second consecutive year that 1-hour ozone values remained below the old standard. As recent as 2002, New Jersey recorded 16 days above this old 1-hour standard.

Figure 8
Highest and Second Highest Daily Ozone 1-Hour Averages

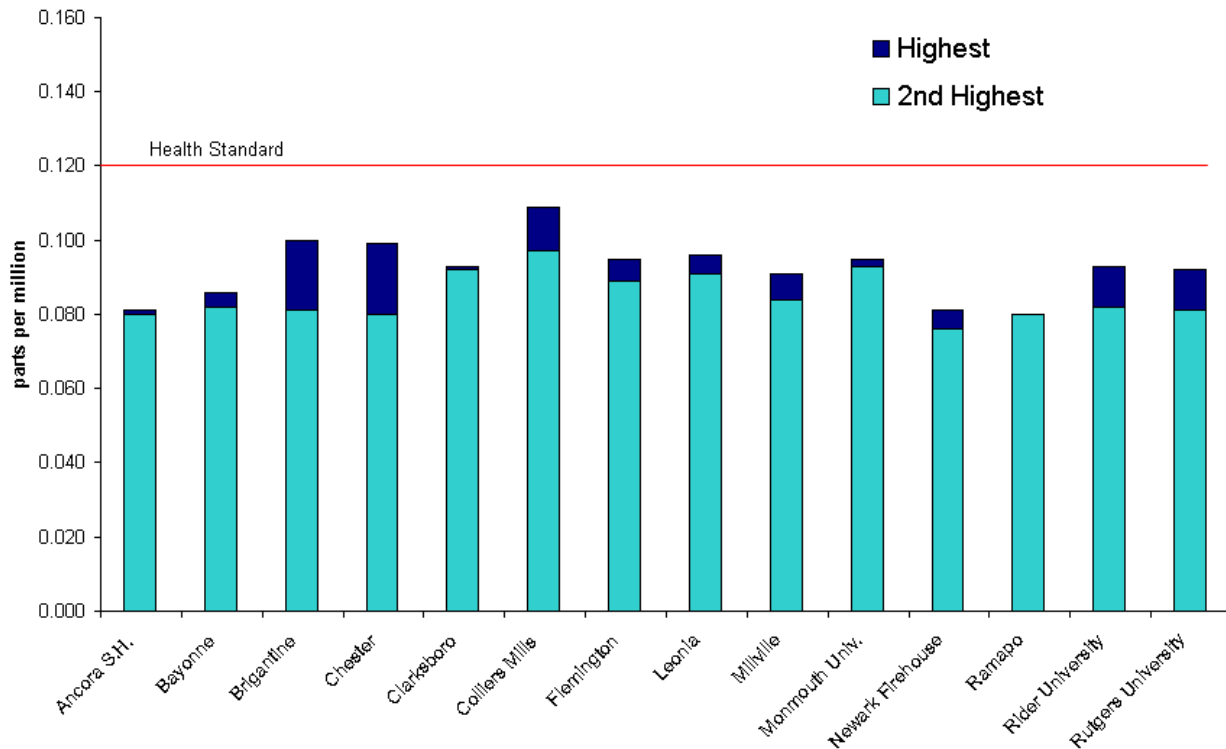


Table 3
Ozone Data – 2009
1-Hour Averages

Parts Per Million (ppm)

Old 1-hour standard is 0.12 ppm

Monitoring Site	1-hr Max	2nd Highest 1-hr Max	4th Highest 1-hour Average 2007-2009	# of days with 1-hour Averages above 0.12ppm
Ancora S.H.	.081	.080	.101	0
Bayonne	.086	.082	.108	0
Brigantine	.100	.081	.089	0
Chester	.099	.080	.101	0
Clarksboro	.093	.092	.107	0
Colliers Mills	.109	.097	.105	0
Flemington	.095	.089	.106	0
Leonia	.096	.091	.102 (3 rd Highest)	0
Millville	.091	.084	.091	0
Monmouth Univ.	.095	.093	.105	0
Newark Firehouse	.081	.076	.076 (2 nd Highest)	0
Ramapo	.080	.080	.097	0
Rider University	.093	.082	.111	0
Rutgers University	.082	.081	.108	0
Statewide	.109	.100		0

SUMMARY OF 2009 OZONE DATA RELATIVE TO THE 8-HOUR STANDARD

Only 8 of the 14 monitoring sites that were operated during the 2009 ozone season recorded levels above the 8-hour standard of 0.075 ppm. The highest 8-hour concentration recorded was 0.085 at Colliers Mills on August 17th. Design values for the 8-hour standard were above the standard at 13 of 14 sites, indicating that the ozone standard is being violated throughout almost all of New Jersey.

Figure 9
Ozone Design Values for 2007-2009

3 Year Average of the 4th Highest 8-Hour Value

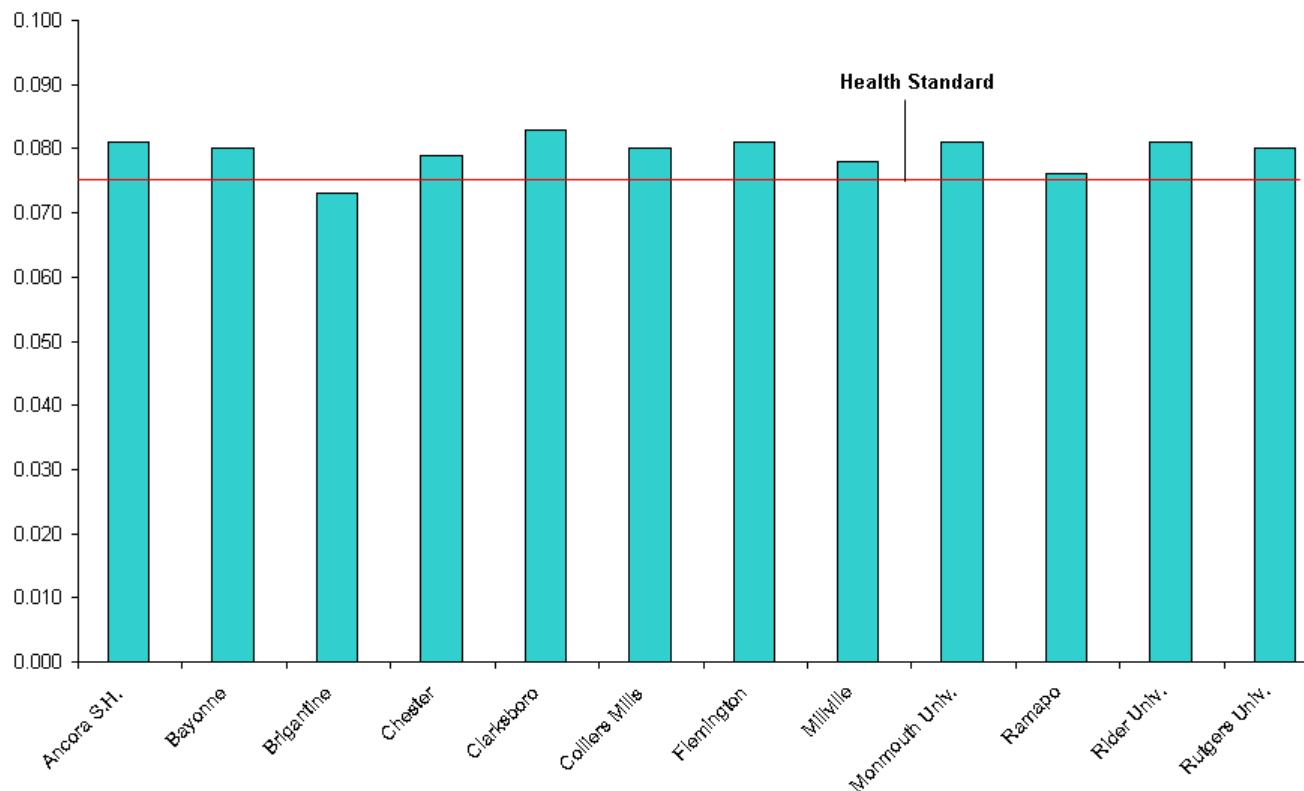


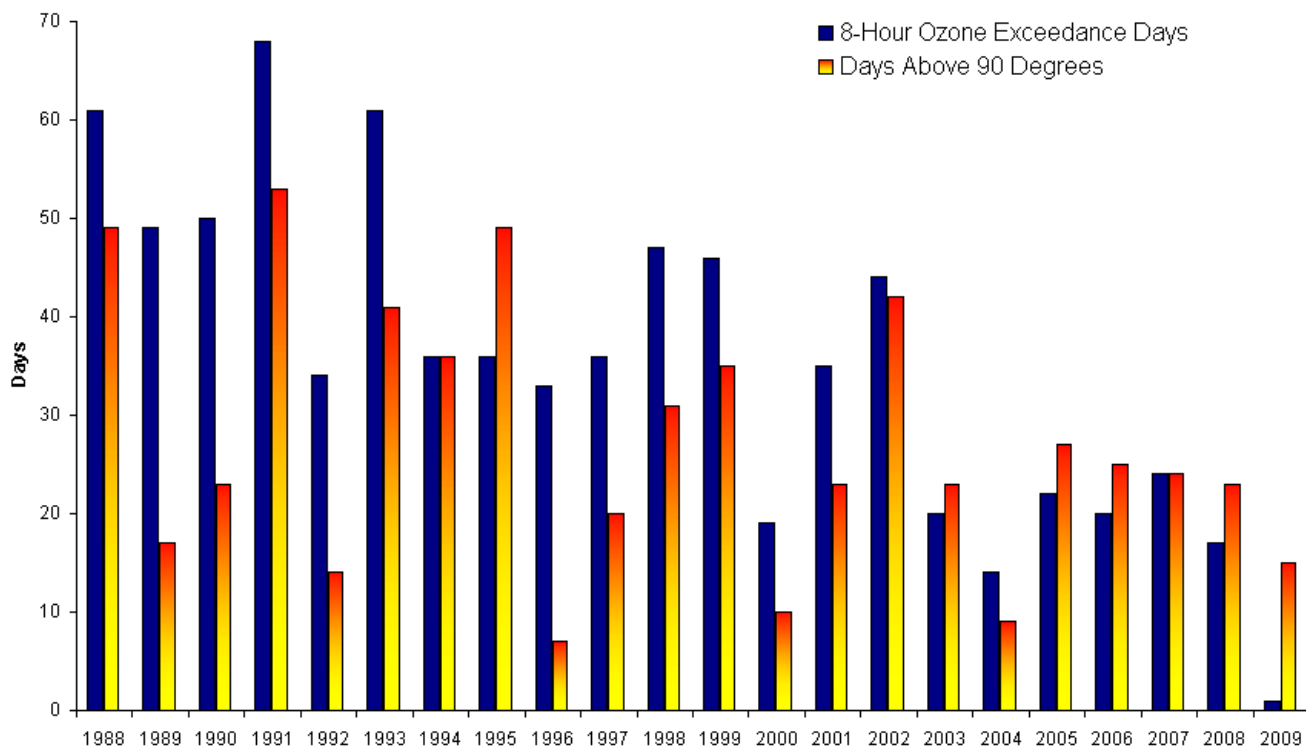
Table 4
Ozone Data – 2009
8-Hour Averages
Parts Per Million (ppm)

8-hour standard is 0.075 ppm

Monitoring Site	1 st Highest	2 nd Highest	3 rd Highest	4 th Highest	Avg. of 4 th Highest 8-hour Averages 2007-2009	# of days with 8-hour above 0.075ppm
Ancora S.H.	.075	.074	.073	.071	.081	0
Bayonne	.074	.072	.072	.069	.080	0
Brigantine	.080	.072	.071	.071	.073	1
Chester	.083	.070	.069	.068	.079	1
Clarksboro	.076	.074	.072	.071	.083	1
Colliers Mills	.085	.080	.072	.071	.080	2
Flemington	.081	.076	.072	.070	.081	2
Leonia	.078	.078	.073	.072	Insufficient data	2
Millville	.079	.075	.073	.072	.078	1
Monmouth Univ.	.078	.078	.073	.072	.081	2
Newark Firehouse	.072	.068	.064	.064	Insufficient data	0
Ramapo	.075	.072	.069	.069	.078	0
Rider University	.073	.073	.072	.071	.081	0
Rutgers University	.072	.068	.067	.067	.080	0
Statewide	.085	.083	.080	.080	.092	9

Figure 10

**Number of Days 8-Hour Ozone Standard was Exceeded and
Number of Days Above 90 Degrees in New Jersey 1988 - 2009**
{Using 8-Hour 0.08 ppm standard (old) across entire time period.}



ACCOUNTING FOR THE INFLUENCE OF WEATHER

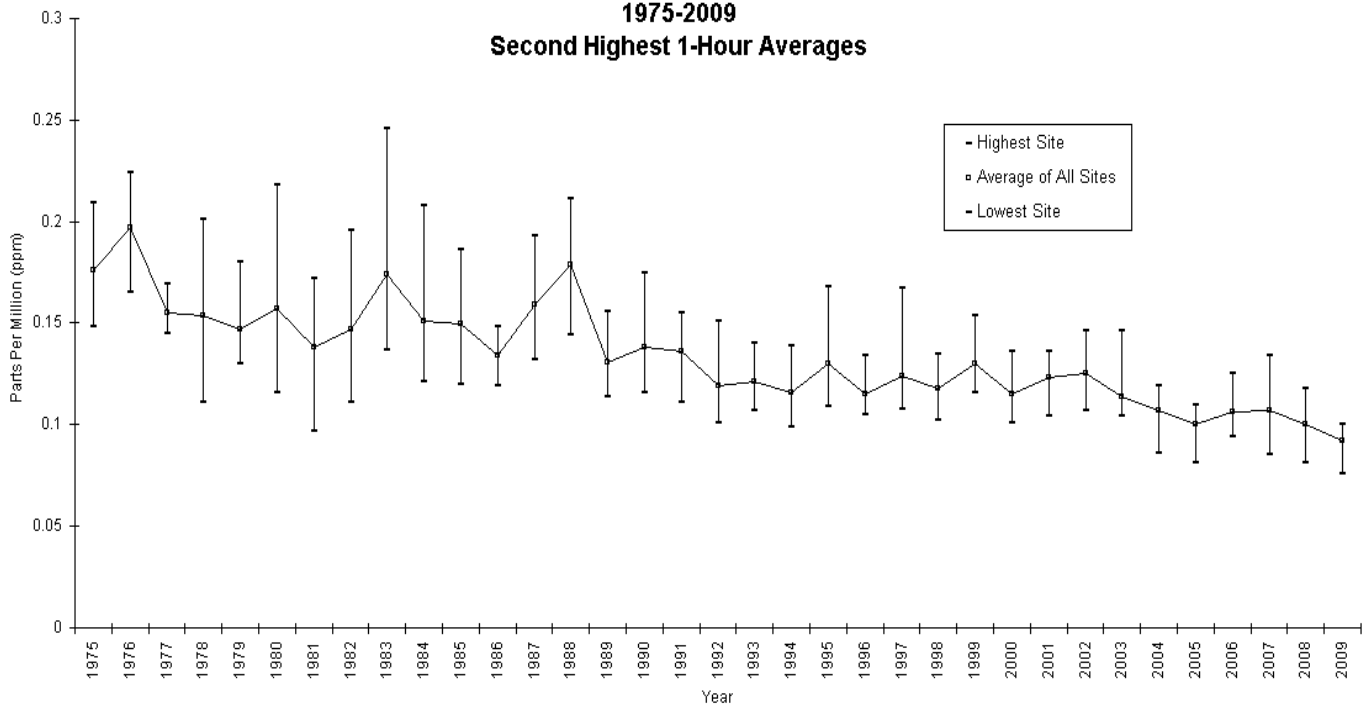
Trends in ground level ozone are influenced by many factors including weather conditions, transport, growth, and the state of the economy, in addition to changes brought about by regulatory control measures. Of these factors, weather probably has the most profound effect on year to year variations in ozone levels. Several methods have been developed to try to account for the effect of weather on ozone levels so that the change due to emissions could be isolated. While none of these methods are completely successful they do show that over the long term, real reductions in ozone levels have been achieved. A simple way of

showing the changing effect of weather on ozone is shown above in Figure 10. The number of days each year on which the ambient temperature was 90 degrees or greater is shown next to the number of days the ozone standard was exceeded. In the earliest years shown (1988-1993) there are significantly more days with high ozone than days above 90 degrees. But this pattern gradually changes and for the most recent years there are more "hot" days than "ozone" days. This is an indication that on the days when conditions are suitable for ozone formation, unhealthy levels are being reached less frequently.

OZONE TRENDS

The primary focus of efforts to reduce concentrations of ground-level ozone in New Jersey has been on reducing emissions of volatile organic compounds (VOCs). Studies have shown that such an approach should lower peak ozone concentrations, and it does appear to have been effective in achieving that goal. Maximum 1-hour concentrations have not exceeded 0.200 ppm since 1988 and the last time levels above 0.180 ppm were recorded was in 1990 (Figure 11). Improvements have leveled off in recent years, especially with respect to maximum 8-hour average concentrations. Significant further improvements will require reductions in both VOCs and NO_x. The NO_x reductions will have to be achieved over a very large region of the country because levels in New Jersey are dependent on emissions from upwind sources.

Figure 11
Ozone Concentrations in New Jersey
1975-2009
Second Highest 1-Hour Averages



OZONE NON-ATTAINMENT AREAS IN NEW JERSEY

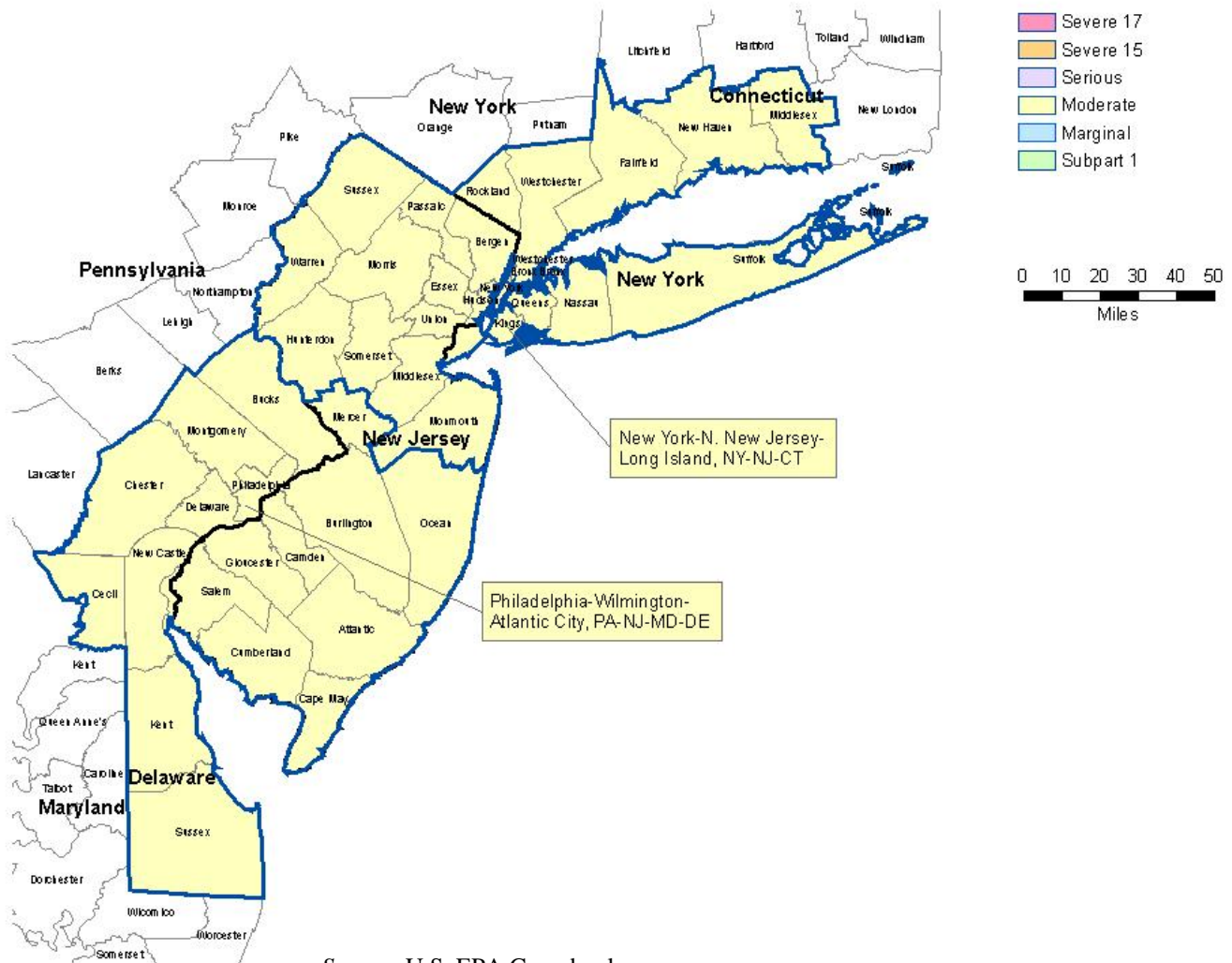
The Clean Air Act requires that all areas of the country be evaluated and then classified as attainment or non-attainment areas for each of the National Ambient Air Quality Standards. Areas can also be found to be “unclassifiable” under certain circumstances. The 1990 amendments to the act required that areas be further classified based on the severity of non-attainment. The classifications range from “Marginal” to “Extreme” and are based on “design values”. The design value is the value that actually determines whether an area meets the standard. For the 8-hour ozone standard for example, the design value is the average of the fourth highest daily maximum 8-hour average concentration recorded each year for three years.

Their classification with respect to the 8-hour standard is shown in Figure 12 below. The entire state of New Jersey is in non-attainment and is classified as being “Moderate.” A “Moderate” classification is applied when an area has a design value from 0.092 ppm to 0.106 ppm.

Figure 12

New Jersey

8-hour Ozone Nonattainment Areas in Blue Border



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